

STUDIES IN THE QUINONE FIELD

XLI. Synthesis of Derivatives of 1-Benzyl-5-hydroxyindole*

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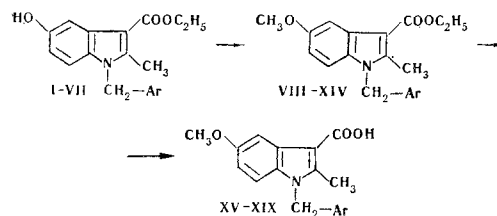
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The condensation of p-benzoquinone with substituted N-benzyl-β-aminocrotonic esters has been studied. A series of derivatives of 1-benzyl-5-hydroxyindole containing various substituents in the benzene nucleus of the benzyl radical has been obtained.

The condensation of quinones with β-aminocrotonic esters is the most effective method for the synthesis of derivatives of 5-hydroxyindole. This reaction, which has been developed in recent years by one of us together with his co-workers [1], has enabled a large number of compounds possessing high biological activity to be obtained [2]. Furthermore, in one of our investigations [3] this reaction was used in an original synthesis of the preparation BAS, which is of considerable interest as a serotonin antagonist [4-6].

In the present paper the condensation of p-benzoquinone with substituted N-benzyl-β-aminocrotonic esters containing various substituents in the benzene nucleus of the benzyl radical, which leads to the formation of 1-benzyl-3-ethoxycarbonyl-5-hydroxy-2-methylindoles (I-VII) is studied. The substituted N-benzyl-β-aminocrotonic esters obtained from acetoacetic ester and nuclear-substituted benzylamines [7-9] were not subjected to purification. The 1-benzyl-3-ethoxycarbonyl-5-hydroxy-2-methylindole derivatives (I-VII) were converted by the action of dimethyl sulfate in alkaline solution into the corresponding methoxy derivatives (VIII-XIV). The latter are stable to hydrolysis under the usual conditions. By saponifying the ethoxycarbonyl group in some of them by fusion with an excess of caustic soda at 180-220° C, a num-

ber of indole-3-carboxylic acids (XV-XIX) have been obtained.


 I, VIII, XV Ar = p-CH₃C₆H₄;

 II, IX, XVI Ar = o-CH₃OC₆H₄;

 III, X, XVII Ar = p-CH₃OC₆H₄;

 IV, XI Ar = p-BrC₆H₄;

 V, XII Ar = p-NO₂C₆H₄;

 VI, XIII, XVIII Ar = 2',4'-(CH₃)₂C₆H₃;

 VII, XIV, XIX Ar = 3',4'-(CH₃O)₂C₆H₃

EXPERIMENTAL

3-Ethoxycarbonyl-5-hydroxy-2-methyl-1-(p-methylbenzyl)indole. (I). A mixture of 24.2 g (0.2 mole) of p-methylbenzylamine, 26 g (0.2 mole) of acetoacetic ester, and one drop of concentrated hydrochloric acid was heated in the water bath for 1 hr and treated with 80 ml of dichloroethane, and the water liberated was distilled off in the form of the azeotropic mixture with dichloroethane (60 ml). After 25-30 min, a solution of 20.4 g of p-benzoquinone in 220 ml of dichloroethane was added to the reaction mixture without the cessation of the distillation of the azeotropic mixture (a total of 230-250 ml was distilled off). The reaction mixture was left overnight and the precipitate was filtered off and washed with methanol. Yield 13.8 g (22%), mp 231-233° C (from a mixture of ethyl acetate and dioxane). Found, %: C 74.11, 74.41; H 6.62, 6.87. Calculated for C₂₀H₂₁NO₃, %: C 74.28; H 6.55. Compounds II-VII were obtained similarly (Table 1).

*For part XL, see [1].

Table 1

Derivatives of 1-Benzyl-3-ethoxycarbonyl-5-hydroxy-2-methylindole

Compound	Mp, °C (solvent for crystallization)	Empirical formula	Found, %			Calculated, %			Yield, %
			C	H	N	C	H	N	
II	203-204 (ethyl acetate)	C ₂₀ H ₂₁ NO ₄	70.84 70.72	6.13 6.26	4.37 4.28	70.78	6.26	4.13	21
III	214-215 (dioxane)	C ₂₀ H ₂₁ NO ₄	70.82 70.62	6.54 6.54	—	70.78	6.26	—	29
IV	204-206 (dioxane)	C ₁₉ H ₁₈ BrNO ₃	58.82 59.04	4.88 5.10	—	58.78	4.77	—	16
V	222-224 (dimethylformamide)	C ₁₉ H ₁₈ N ₂ O ₅	64.51 64.63	5.23 5.25	7.66 7.85	64.40	5.12	7.91	23
VI	236-237 (dimethylformamide)	C ₂₁ H ₂₃ NO ₃	74.65 74.51	7.09 6.91	4.38 4.57	74.75	6.87	4.15	26
VII	154-155 (dioxane)	C ₂₁ H ₂₃ NO ₅	68.33 68.33	6.47 6.37	—	68.27	6.27	—	18

Table 2

Derivatives of 1-Benzyl-3-ethoxycarbonyl-5-methoxy-2-methylindole

Compound	Mp, °C (solvent for crystallization)	Empirical formula	Found, %			Calculated, %			Yield, %
			C	H	N	C	H	N	
IX	105—107 (isopropanol)	C ₂₁ H ₂₃ NO ₄	—	—	3.98 4.15	—	—	3.96	91
X	108—109 (isopropanol)	C ₂₁ H ₂₃ NO ₄	71.68 71.50	6.84 6.79	—	71.37	6.56	—	86
XI	146—148 (dioxane)	C ₂₀ H ₂₀ BrNO ₃	59.62 59.77	5.10 5.22	3.63 3.51	59.70	5.01	3.48	72
XII	153—155 (dioxane)	C ₂₀ H ₂₀ N ₂ O ₅	65.02 65.05	5.67 5.51	7.70 7.90	65.19	5.48	7.60	84
XIII	114—115 (ethanol)	C ₂₂ H ₂₅ NO ₃	74.89 74.71	7.07 7.16	3.87 4.10	75.19	7.14	3.99	83
XIV	119—121 (ethanol)	C ₂₂ H ₂₅ NO ₅	69.32 69.43	6.71 6.63	—	68.91	6.57	—	77

Table 3

Derivatives of 1-Benzyl-5-methoxy-2-methylindole-3-carboxylic Acid

Compound	Mp (decomp.), °C (solvent for crystallization)	Empirical formula	Found, %		Calculated, %		Yield, %
			C	H	C	H	
XVI	210—213 (ethyl acetate—dioxane)	C ₁₉ H ₁₉ NO ₄	70.36 70.07	5.73 5.69	70.13	5.88	71
XVII	195—197 (dimethylformamide)	C ₁₉ H ₁₉ NO ₄	70.21 70.22	6.01 6.05	70.13	5.88	90
XVIII	201—203 (dioxane)	C ₂₀ H ₂₁ NO ₃	74.51 74.32	6.86 6.72	74.27	6.55	86
XIX	179—181 (dioxane—ethanol)	C ₂₀ H ₂₁ NO ₅	67.93 67.81	6.28 6.29	67.58	5.96	67

3-Ethoxycarbonyl-5-methoxy-2-methyl-1-(p-methylbenzyl)indole (VIII). To a solution of 3.53 g (0.011 mole) of I in a mixture of 12 ml of 2 N NaOH and 20 ml of dioxane 10-15 minutes after its preparation was added 1.5 ml of freshly-distilled dimethyl sulfate, and the mixture was stirred at 20° C for 1 hr and poured into 200 ml of water. The oil that separated out crystallized on being rubbed with a rod. The crystals were filtered off and washed with water. Yield 3.27 g (87%), mp 99-100° C (from dioxane). Found, %: C 74.92, 74.72; H 6.88, 7.03. Calculated for $C_{21}H_{23}NO_3$, %: C 74.75; H 6.87.

The other 5-methoxyindoles IX-XIV were obtained similarly (Table 2).

5-Methoxy-2-methyl-1-(p-methylbenzyl)indole-3-carboxylic acid (XV). To a melt of 6.4 g (0.16 mole) of caustic soda and 1.2 ml of water was added 4.42 g (0.013 mole) of VIII. The reaction mixture was stirred vigorously at 200-220° C for 10-15 min. The cooled melt was dissolved with heating in 70 ml of water, the hot solution was filtered, and the filtrate was cooled and acidified with dilute hydrochloric acid. Yield 3.63 g (73%); mp 212-214° C (decomp., from dioxane). Found, %: N 4.47, 4.42. Calculated for $C_{19}H_{19}NO_3$, %: N 4.52.

The other indolecarboxylic acids (XVI-XIX) were obtained similarly (Table 3).

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